

SYNEILESINE, A NEW PYRROLIZIDINE ALKALOID FROM SYNEILEISIS PALMATA¹

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In the course of our continuing chemical study on pyrrolizidine alkaloids in crude drugs², we have isolated a new pyrrolizidine alkaloid, together with two unknown alkaloids, from the fresh and dried roots of Syneileisis palmata Maxim. (Japanese name: Yaburegasa, Compositae) which was collected near the Sagami-ko, Kanagawa-ken, Japan in 1973.

The present paper deals with the structure determination of a new secopyrrolizidine alkaloid, named syneilesine(I) which has highly cytotoxic activity³.

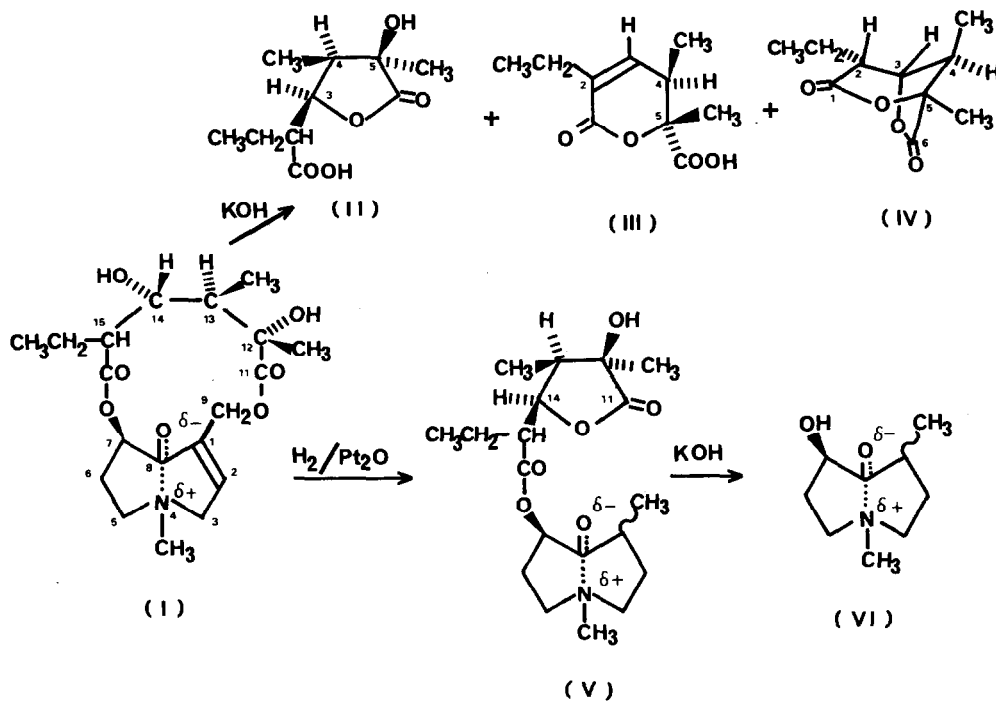
The crude alkaloid extracted from the MeOH ext. of the dried roots (5.08 kg) was chromatographed on silica gel column and eluted with CHCl₃-MeOH-NH₄OH solvent system to yield syneilesine(I), colorless needles (from petroleum b.p. 60-80°), 0.762 g (0.015%), m.p. 195°, CD [θ]_{max}^{25°} (MeOH); +30500 (232nm), +43500 (275), C₁₉H₂₉O₇N, IR ν _{max}^{KBr}; 3500cm⁻¹, 3450, 1735, 1720. ¹³C-NMR spectrum shows 19 detectable signals of carbon, of which signals at δ ; 134.2ppm, 136.0, 171.4, 176.7 and 189.4 were easily assigned⁴. The signals at 171.4ppm and 176.7 indicate the presence of two ester carbonyl carbons⁵ which were also ascertained by IR spectral absorption at 1720cm⁻¹ and 1735. The signals at 134.2ppm, 136.0 and 189.4 are responsible for α,β-unsaturated carbonyl group. A little high field shift of the signal of carbonyl carbon at 189.4ppm, relative to the ordinary α,β-unsaturated carbonyl carbons⁶, should be caused from the influences of transannular interactions of the nitrogen atom in the pyrrolizidine nucleus⁷. Proton NMR spectral pattern of the alkaloid also shows a typical macrocyclic secopyrrolizidine alkaloid; a singlet at δ 2.07ppm corresponds to CH₃-N< at N-4, two broad singlets at 6.05ppm and 5.02 to the olefinic proton at C-2 and >CH-OCO- at C-7, respectively. The complicated peaks at 2.00ppm to 3.70 are due to the methylene protons at C-3, C-5 and C-6 in secopyrrolizidine nucleus. The signals of the geminal protons at C-9 are represented as a pair of doublets at 5.50ppm and 4.80 (J=11.5Hz each other).

From the appreciable difference of the shift ($\delta_{\text{H}}=0.70\text{ppm}$) between the geminal protons and the coupling constant ($J=11.5\text{Hz}$), syneilesine is classified as a characteristic 12-membered macrocyclic secopyrrolizidine alkaloid⁸. Other assignable signals are at 0.91ppm (3H, t, $J=7.5\text{Hz}$) for $\text{CH}_3\text{-CH}_2\text{-}$ and 1.50 (3H, d, $J=7.5\text{Hz}$) for $\text{CH}_3\text{-CH<}$, respectively.

High resolution mass spectrometric studies on this alkaloid showed that the fragment ions at m/e ; 168 ($\text{C}_9\text{H}_{14}\text{O}_2\text{N}$), 152 ($\text{C}_9\text{H}_{14}\text{ON}$) and 151 ($\text{C}_9\text{H}_{13}\text{ON}$) arose from the secopyrrolizidine moiety⁹. The significant fragment ions at m/e ; 339 ($\text{C}_{18}\text{H}_{29}\text{O}_5\text{N}$) and 266 ($\text{C}_{14}\text{H}_{20}\text{O}_4\text{N}$) indicate that the two hydroxyl groups are located at C-12 and C-14 in the necic acid moiety.

Hydrolysis of syneilesine with 10% KOH in EtOH gave three new lactones, such as syneilesinolide-A (γ -lactone), m.p. 134°, $\text{CD}[\theta]_{\text{max}}^{15^\circ}(\text{MeOH})$; +3700 (215nm), $\text{C}_{10}\text{H}_{16}\text{O}_5$, $\text{IR} \nu_{\text{max}}^{\text{KBr}}$; 3400 cm^{-1} , 1790, 1700, $\text{PMR}(\text{CDCl}_3) \delta$; 1.04ppm (3H, t, $J=7.5\text{Hz}$), 1.00 (3H, d, $J=6.2$), 1.52 (3H, s), 4.58 (1H, d-d, $J=10.0$ and 5.2), syneilesinolide-B (α, β -unsaturated δ -lactone), m.p. 121°, $\text{CD}[\theta]_{\text{max}}^{15^\circ}(\text{MeOH})$; -82500 (203nm), -61000 (223), -6600 (257), $\text{C}_{10}\text{H}_{14}\text{O}_4$, $\text{IR} \nu_{\text{max}}^{\text{KBr}}$; 3300 cm^{-1} , 1740, 1690, $\text{PMR}(\text{CDCl}_3) \delta$; 0.98ppm (3H, d, $J=7.5\text{Hz}$), 1.09 (3H, t, $J=8.0$), 1.58 (3H, s), 2.30 (2H, broad q, $J=8.0$), 2.88 (1H, d-q, $J=7.5$), 6.58 (1H, broad d, $J=7.5$) and syneilesinolide-C (γ, δ -dilactone), m.p. 86°, $\text{CD}[\theta]_{\text{max}}^{15^\circ}(\text{MeOH})$; +9800 (210nm), $\text{C}_{10}\text{H}_{14}\text{O}_4$, $\text{IR} \nu_{\text{max}}^{\text{KBr}}$; 1790 cm^{-1} , 1745, $\text{PMR}(\text{CDCl}_3) \delta$; 0.92ppm (3H, t, $J=7.5\text{Hz}$), 0.98 (3H, d, $J=8.0$), 1.50 (3H, s), 4.75 (1H, d-d, $J=5.4$ and 3.0), 1.60-2.60 (3H, complicated peaks). Hydrogenation of syneilesinolide-B gave dihydrosyneilesinolide-B, m.p. 85°, $\text{C}_{10}\text{H}_{16}\text{O}_4$, $\text{CD}[\theta]_{\text{max}}^{20^\circ}(\text{MeOH})$; -2400 (238nm), whose negative Cotton effect at 238nm indicates that the configuration at C-5 in dihydrosyneilesinolide-B is the same with that at C-2 in (2R)-dihydroseneic acid¹⁰. From IR, NMR, Mass and CD spectroscopic studies, syneilesinolide-A, syneilesinolide-B and syneilesinolide-C were assumed to be (3R), (4R), (5R)-2-ethyl-5-hydroxy-4,5-dimethylhexanoic acid-6,3-olide (II), (4R), (5R)-5-carbohydroxy-2-ethyl-4,5-dimethyl-2-pentene-5-olide (III) and (2R), (3R), (4R), (5R)-2-ethyl-4,5-dimethylhexane-1,5:6,3-diolide (IV), respectively.

On the other hand, hydrogenolysis of syneilesine was carried out with Adam's catalyst in dil. HCl solution to give dihydrosyneoxyneilesine-11,14-olide (V), m.p. 109.5°, $\text{CD}[\theta]_{\text{max}}^{16^\circ}(\text{MeOH})$; -17500 (230nm), $\text{C}_{19}\text{H}_{31}\text{O}_6\text{N}$, $\text{IR} \nu_{\text{max}}^{\text{KBr}}$; 3450 cm^{-1} , 1770, 1740, 1620, which was hydrolyzed to give the necic acid, syneilesinolide-A, syneilesinolide-B and a small amount of syneilesinolide-C and the necine, dihydrosyneoxytonecine (VI)⁷, whose hydrochloride was identical with the authentic sample by mixed m.p., $[\alpha]_{\text{D}}$ and IR spectrum.



The structure of the basic moiety and the acidic moiety and the formation of dihydrodesoxytotonine-11,14-olide leads to the conclusion that the structure of synephrine is (12R), (13R), (14R)-15-ethyl-12,14-dihydroxy-4,12,13-trimethyl-8-oxo-4,8-seco-senec-1-enine(I).

The structural studies and cytotoxic bioassay of other alkaloids are now in progress.

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* All melting points are uncorrected and the molecular formulae were measured by high resolution mass spectrometer and the analytical values were in good agreement with the calculated values.

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